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The concept of expressing acidity as the negative logarithm of the hydrogen ion concentration was defined and termed pH in the beginning of the 20th century. The general usefulness of the pH concept for life science was recognized and later gained importance to analytical research. Reports on results of pH measurements from living skin established the term acid mantle - the skin's own protective shield that maintains a naturally acid pH. It is invisible to the eye but crucial to the overall wellbeing of skin. Chronic

alkalization can throw this acid mantle out of balance, leading to inflammation, dermatitis, and atopic skin diseases. It is therefore no surprise, that skin pH shifts have been observed in various skin pathologies. It is also obvious that the pH in topically applied preparations may play an important role. Optimal pH and buffer capacity within topical preparations not only support stability of active ingredients and auxiliary materials, but may also increase absorption of the non-ionized species of an acidic or a basic active ingredient. They may even open up opportunities to modify and "correct" skin pH and hence accelerate barrier recovery and maintain or enhance barrier integrity. Further efforts are needed to standardize and improve pH measurements in biological media or pharmaceutical/cosmetic vehicles to increase and ensure quality, comparability, and relevance of research data. In this volume, we present a unique collection of papers that address past, present and future issues of the pH of healthy and diseased skin. It is hoped that this collection will foster future efforts in clinical and experimental skin research. "PH and Brain Function offers thorough coverage of this increasingly important area of research, beginning with the fundamental concepts, which include methodological and theoretical issues such as the measurement of pH and the concept of pH in neurobiology. It explores aspects of regulation and modulation of intracellular pH in brain cells, surveys the changes in pH that occur with neural activity and how these changes affect neural activity, and discusses the role of pH in the pathophysiology of neurological diseases." "pH and Brain Function is an important resource for researchers in all areas of neuroscience as well as cell biology and physiology." --Book Jacket. Alloy X-750 condition HTH stress corrosion crack growth rate (SCCGR) tests have been conducted at 360 C (680 F) with 50 cc/kg hydrogen as a function of coolant pH. Results indicate no appreciable influence of pH on crack growth in the pH (at 360 C) range of (almost equal to) 6.2 to 8.7, consistent with previous alloy 600 findings. These intermediate pH results suggest that pH is not a key variable which must be accounted for when modeling pressurized water reactor (PWR) primary water SCC. In this study, however, a nearly three fold reduction in X-750 crack growth rate was observed in reduced pH environments (pH 3.8 through HCl addition and pH 4-5.3 through H<sub>2</sub>SO<sub>4</sub> addition). Crack growth rates did not directly correlate with corrosion film thickness. In fact, 10x thicker corrosion films were observed in the reduced pH environments. Considerable variation was noted in concentration of the seven major minerals namely Ca, Mg, P, acid, K, Na, and Cl in soluble phase of milk each at 6 different PH levels. Soluble calcium was found decreasing when the milk PH has been increased. Soluble magnesium decreased with increase in PH. ON increasing the milk PH soluble P showed small irregular variations. When PH was lowered slight increase in citric acid value was noticed potassium was unchanged when milk PH was changed to 5,6 and 7 but gradually decreased on increasing the milk PH to 8 and 9. It as observed that soluble forms of these minerals contributed to over 81% towards PH of buffaloes milk and rest 19% was contribution from other ingredients it was observed that observed that heat coagulation time (HCT) decreased sharply to 0.30±0.040 and further to 0.029±0.005 minutes at PH 6 and 5 respectively rise in PH of milk to 7,8 and 9 increased HCT to 92.90±10.976, 156.08±7.956 and 166.40±8.912 minutes respectively. The aims of this study were to detect the amount of fluoride released from the tested materials into three different storage solutions and to determine the change in the surface microhardness after storage. The amount of fluoride released from the three materials was significantly higher in acidic solutions than in neutral solutions. The Compomer released the highest amount of fluoride into the three storage solutions, while the least fluoride release was from the resin-modified glass-ionomer. The surface microhardness of the three tested materials reduced significantly after storage in neutral and acidic solutions compared to dry stored specimens. In acidic solutions, the surface microhardness was significantly lower than in neutral solutions. The Compomer showed the highest microhardness number values. A unified overview of the dynamical properties of water and its unique and diverse role in biological and chemical processes. Skin surface pH -- Franz diffusion cells -- Platinum group metals -- Rhodium -- Ionisation -- Oxidation -- Vel oppervlak pH -- Franz diffusie selle -- Platinum groep metale -- Rodium -- Ionisatie -- Oksidasie. Microbial responses to acidic and alkaline pH are important in many areas of bacteriology. For example, the mechanisms of resistance to acidic pH are important in the understanding of the passage of human pathogens through the acid of the stomach; and an understanding of microbial degradation of alkaline industrial waste is important for the environment. Bringing together contributions from an international and interdisciplinary group of experts working on the many aspects of bacterial cellular responses to pH, this stimulating volume draws together new and innovative work in this area. It delineates both similarities and differences between mechanisms of tolerance and response, providing readers with an invaluable resource on the subject.

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